

Discovery and Significance of High CH₄ Primary Fluid Inclusions in Reservoir Volcanic Rocks of the Songliao Basin, NE China

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Abstract: Comparing compositions of the fluid inclusions in volcanic rocks to the contents and isotopes of the gases in corresponding volcanic reservoirs using microthermometry, Raman microspectroscopy and mass spectrum analysis, we found that: (1) up to 82 mole% methane exists in the primary inclusions hosted in the reservoir volcanic rocks; (2) high CH₄ inclusions recognized in the volcanic rocks correspond to CH₄-bearing CO₂ reservoirs that are rich in helium and with a high ³He/⁴He ratio and which show reversed order of $\delta^{13}\text{C}$ in alkane; (3) in gas reservoirs of such abiogenic methane (>80%) and a mix of CH₄ and CO₂, the enclosed content of CH₄ in the volcanic inclusions is usually below 42 mole%, and the reversed order of $\delta^{13}\text{C}$ in alkane is sometimes irregular in the corresponding gas pools; (4) a glassy inclusion with a homogeneous temperature over 900°C also contains a small portion of CH₄ although predominantly CO₂. This affinity between gas pool and content of inclusion in the same volcanic reservoirs demonstrates that magma-originated gases, both CH₄ and CO₂, have contributed significantly to the corresponding gas pools and that the assumed hydrocarbon budget of the bulk earth might be much larger than conventionally supposed.

Key words: Songliao Basin, reservoir volcanic rocks, fluid inclusions, methane (CH₄), carbon dioxide (CO₂), abiogenic origin natural gas

1 Introduction

Abiogenic methane has been proven in deep-seated fluid inclusions and can also be generated by several processes such as respeciation of magmatic volatiles (Kelley, 1996), the Fischer-Tropsch synthesis (Salvi and William-Jones, 1997), and geothermal sources (Sheppard, 1981). However, whether these abiogenic hydrocarbons have any commercial value is still in question. One case report is provided by natural gases commercially produced from volcanic rocks in northeastern Japan. Wakita and Sano (1983) claimed, based on ³He/⁴He measurement, that more than one-third of the total methane might have been derived from upper mantle magmatic activity. But Sakata et al. (1989) thought, based on geochemical analysis, that no more than 0.1% of the total methane can be attributed to a magmatic origin. A similar debate has also taken place for the Songliao Basin of NE China.

In recent years there has been burgeoning interest in and debate about the abiogenic methane accumulated in commercial quantities in the Mesozoic volcanic reservoirs of the Songliao Basin. People in favor of an inorganic origin for the natural gas (e.g. Guo et al., 1994, 1997) provided the following evidence: (1) reservoirs are controlled by deep-seated faults and show close affinity to volcanism; (2) the isotopically heavy CH₄ ($\delta^{13}\text{C} > -30\text{‰}$) is similar to that typical of abiogenic methane enclosed in mantle-derived inclusions in the basalt of the region (Liu et al., 1998; Wang et al., 2000); (3) the isotopic composition of co-existing helium is ca. 3 times the R_a (where R_a denotes the atmospheric ³He/⁴He ratio of 1.39×10^{-6}), presenting a mantle-derived signature (Xu et al., 1995); (4) the reversed orders of $\delta^{13}\text{C}$ in alkane ($\delta^{13}\text{C}_1 > \delta^{13}\text{C}_2 > \delta^{13}\text{C}_3 > \delta^{13}\text{C}_4$) suggest abiogenic contributions (Dai et al., 2003). However, others have argued that methane-dominant ($\text{C}_1/\text{C}_{2+3} > 100$) coal gas can also have $\delta^{13}\text{C}_{\text{CH}_4}$ values above -30‰ (Dai, 1993) and so noble gases like helium may not originate from the same source as that of the co-existing

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Table 1 Analytical data for selected inclusions in quartz of the representative reservoir volcanic rocks of the Songliao Basin

Sample	FS9-2		FS9-1				ZS6-1				SHS2		SOS1			ZS5	SiS1	FS7		
Inclusion No.	1	4*	0	1	2	3	4	1	2	3	4	6	1*	1	2*	3*	1*	1*	1*	
<i>Th</i> (°C)	375		368	395	315	312	295	386	298	305	293	306		377						>900
Filling degree	0.1	0.0	0.1	0.5	0.7	0.7	0.7	0.3	0.7	0.7	0.7	0.6	0.0	0.3	0.0	0.0	0.0	0.0	0.0	0.7
<i>T_{mCO2}</i> (°C)	-64.2	-69.2	-66.1	-63.5	-61.0	-66.5	-64.6	-62.7	-56.9	-57.2	-56.9	-56.9	-62.5	-62.8	-64.0	-59.7	-59.7	-63.2	-57.0	
<i>Th_{CO2}</i> (°C)	-20.0	-45.2	-39.4	-4.3	7.2	-38.5	-30.2	-6.8	8.6	8.0	8.4	8.0	-8.3	-10.5	-12.3	4.8	5.2	-10.3	7.9	
<i>T_{mclath}</i> (°C)	13.9		14.8	12.1	10.9	14.8	13.6	12.9	8.2	8.1	8.4	8.3		13.2						
Salinity (%)	5.0±		5.0±	4.4±	4.0±	5.0±	4.9±	4.0±	5.0	4.4	4.6	4.7		4.8±						
Dv	0.64		0.53	0.54	0.73	0.45	0.63	0.61	0.75	0.69	0.72	0.70		0.75						
DI	1.06		1.09	1.10	1.10	1.08	1.07	1.06	1.07	1.08	1.07	1.07		1.06						
Vapor phase (mole %)	CH ₄ 40.5	82.5	66.1	29.0	16.8	61.9	49.9	27.9	3.7	3.9	1.0	2.5	34.2	30.7	43.4	16.6	14.8	31.8	2.7	
	CO ₂ 59.5	17.5	33.9	71.0	83.2	38.1	50.1	72.1	96.3	96.1	99.0	97.5	65.8	69.3	56.6	83.4	85.2	68.2	97.3	
Bulk comp. (%)	CH ₄ 31.2	82.5	58.8	5.4	1.9	5.9	8.3	11.0	0.6	0.8	0.2	0.3	34.2	14.0	43.4	16.6	14.8	31.8	2.7	
	CO ₂ 52.0	17.5	25.6	15.3	11.3	5.7	4.5	29.5	14.5	17.7	16.2	19.5	65.8	37.6	56.6	83.4	85.2	68.2	97.3	
	H ₂ O 16.8		15.6	79.3	86.8	88.4	87.2	59.5	84.9	81.5	83.6	80.2		48.4						

Notes: (1) Samples with asterisk are monophasic or melt inclusions; (2) symbols as in text, *Th*: temperature of total homogenization of the inclusion (°C); *T_{mCO2}* (*T_{mclath}*): melting temperature of solid CO₂ (clathrate) in the presence of vapor (°C); *Th_{CO2}*: homogenization temperature of the non-aqueous part (°C); DI (Dv): density of the aqueous part (of the non-aqueous part) (g/cm³).

methane. In addition, the reversed orders of $\delta^{13}\text{C}$ in alkane can also be generated by such processes as mixing of genetically different gases, either by evolution of the source rocks themselves or by leaking fractionation within cap rocks (Zheng et al., 1999; Li et al., 1999; Huang, 2000).

Owing to the variety of source rocks and complex geological settings of the Songliao Basin (Wang et al., 1993), most of the reservoirs could be composed of mixed gases from different origins possibly including biodegraded alkanes (Wang et al., 2004). So the contribution of abiogenic alkane to the commercial gas pools is still uncertain. A key point to this problem is to provide direct evidence for abiogenic gases and to indicate their relationship with the commercial gases hosted in the reservoir. The purpose of this paper is to assess the contribution of abiotic methane to the commercial gas reservoirs in the Songliao Basin. Commercial natural gases trapped in volcanic reservoir constitute mobile gases in that they can migrate within the reservoir. In contrast with these exploitable gases, the fluids hosted in minerals of the volcanic rocks are restricted gases. We compare composition and isotopes between the two kinds of gases contained in the same volcanic rocks in order to find the relationship between them. And finally we hope to get a conclusive answer about whether or not these abiogenic gases have contributed significantly to the corresponding commercial gas pools.

2 Geological Setting and Sampling

Situated in the northeast (Fig. 1), the Songliao Basin is the most important oil and gas-bearing basin in China owing to its annual crude oil equivalent production of over 50 million tons. The rocks of the basin are stratigraphically

composed of two different units. The lower part is composed of Upper Jurassic (Oxfordian) to Lower Cretaceous (Aptian) volcanic-pyroclastic-epiclastic successions of up to 3500 m thick (Wang et al., 2002a). The upper part comprises a Lower Cretaceous (Albian) to Upper Cretaceous (Maastrichtian), sedimentary sequence of up to 5700 m thick (Wang et al., 1996). The volcanogenic successions are our main interest in this paper as they host the organic component for the oil.

Natural gases that are above cut-off grade for industry use include hydrocarbons, carbon dioxide and some noble gases like helium. The volume percent of the volcanic and sedimentary reservoirs is about half and half (50:50) in the studied area (Fig. 1). The volcanic rocks cover a wide spectrum from rhyolite to basaltic andesite with ages ranging from late Jurassic to early Cretaceous (Wang et al., 2002b). Typically the volcanic rocks are phenocryst-poor, with porphyritic texture and a complex phenocryst assemblage commonly including quartz, feldspars (sanidine, anorthoclase and plagioclase), biotite, magnetite and ilmenite. The phenocrysts are generally less than 20% in volume of the volcanic rocks and tuffs. Rock names and textures used in the paper follow the classifications of Le Maitre et al. (1989) and MacKenzie et al. (1982). Fluid inclusions seen in doubly polished wafers (200–300 μm in thickness) were carefully selected under the microscope; quartz crystals were selected, including from the rhyolite, dacite and andesite, which are the predominant volcanic rock reservoirs. For each sample we determined fluid inclusion compositions and homogenization temperature, conducted with microthermometry and Raman microspectroscopy (cf. Table 1) after having done thin-section, bulk, trace, and REE element analysis and isotopic dating (e.g. TAS, total alkali versus silica diagram, seen in

Table 2 Bulk natural gas components and isotopes hosted in the reservoir volcanic rocks corresponding to the fluid inclusion-bearing samples shown in Table 1

Well No.	FS9-2	FS9-1	ZS6	ShS2	SoS1	ZS5	SiS1	FS7
Depth (m)	3582	3697	3590	2907	3470	3529	3700	3710
Lithology	Volcanics	Volcanics	Volcanics	Tuff	Volcanics	Tuff	Tuff	Tuff
TAS	Rhyolite	Rhyolite	Rhyolite	Trachydacite	Trachyandesite	Rhyolite	Rhyolite	Rhyolite
Gas pool type	CH ₄ -bearing CO ₂	CH ₄ -bearing CO ₂	CO ₂ -bearing CH ₄	CO ₂ -bearing CH ₄	CO ₂ -bearing CH ₄	CO ₂ -bearing CH ₄	Mixture	Mixture
Production (m ³ /day)	>40000	>1000	>100	>3×10 ⁵	>1000	>200	>20000	>40000
Gas component (mole%)	CH ₄	15.1	10.9	90.5	81.6	96.4	93.4	37.9
	C ₂ ⁺	0.23	0.16	3.37	2.37	0.77	1.98	0.12
	C ₁ /ΣC ₁ ⁺	0.99	0.99	0.96	0.97	0.99	0.98	0.99
	CO ₂	82.2	85.7	2.38	6.98	2.30	0.22	50.6
	He	2.44	2.74	0.06	0.01	0.02	0.03	n.d.
Isotope composition (‰) _{PDB}	CH ₄	-27.11	-27.25	-25.90	-28.71	-26.44	-25.90	-28.00
	C ₂ H ₆	-30.05	n.d.	-28.98	-28.95	-23.44	-28.98	-34.00
	C ₃ H ₈	-30.50	n.d.	-32.00	-35.01	-32.82	-32.00	-34.10
	C ₄ H ₁₀	-32.98	n.d.	n.d.	-37.58	-25.02	n.d.	n.d.
	CO ₂	-4.06	-5.46	n.d.	n.d.	n.d.	n.d.	n.d.
³ He/ ⁴ He (×10 ⁻⁶)		4.2	4.0	n.d.	n.d.	n.d.	n.d.	n.d.

Note: (1) Gas samples were collected by Qijun Hou and analyzed at Daqing Institute of Exploration with mass spectrometry; (2) n.d. indicates not detected. Standard deviation for the carbon isotope analysis is from 0.5 to 0.8.

Table 1 of Wang et al., 2006).

As shown in Fig. 1, commercial reservoirs in the studied area can be classified into four groups according to a combination between reservoir rocks and gas geochemistry. They include: volcanic reservoirs, (a), hosting gases with heavy and reversed carbon isotopes; (b), hosting gases without heavy or reversed carbon isotopes; and sedimentary reservoirs, (c), hosting gases with heavy and reversed carbon isotopes; and (d), hosting gases without heavy or reversed carbon isotopes. We chose samples only from the first group for our study.

Up to now over 50 exploration wells have been drilled through these volcanic reservoirs. We selected eight of them for our study on the basis of their distribution, geology and gas-geochemistry (Dai et al., 1996). Drill cores of volcanic rocks and tuffs were collected from the eight representative gas reservoirs covering the three gas types of CO₂, CH₄ and CO₂-CH₄ mixtures (Table 2). In order to minimize the differences between samples arising from different ages and mineral evolution of the rocks, we chose quartz-bearing Lower Cretaceous (Barremian) rocks (Wang et al., 2002a) and, as noted above, used only quartz crystals for fluid inclusion measurements. Three types of fluid inclusions were recognized including secondary, primary and glassy ones. The latter two are of interest in this work.

3 Analytical Techniques

Microthermometry and Raman microspectroscopy were conducted at the Geofluid Laboratory of Jilin University with a British Renishaw System-1000 Laser Raman Spectrometer equipped with a Leica DMLM (objective ×50) microscope and a Linkam THMS-600 Heating-

Freezing Stage. Micro-Raman analytical conditions were as follows: Mark 514.532 nm Ar⁺ laser generator with a laser power of 20 mw, 1800 g/mm grating, slit width of 20 μm, scanning span 4500–850 cm⁻¹, integrated time 120 s. A charge coupled device (CCD) array detector was employed to detect the optical signal produced by the sample after analysis with an analytical precision of 1 cm⁻¹±. The spectrometer was calibrated with monocrystalline silica (520 cm⁻¹) before determination. The Heating-Freezing Stage was calibrated before microthermometric determinations at temperatures of -56.6°C, 0°C and 374.0°C with man-made inclusions composed of 25% H₂O–75 mole% CO₂ and pure water. The analytical precision is 0.1°C ± at low temperature (<30°C) and 2°C ± at high temperature.

The Micro-Raman quantitative data of the fluid inclusions were calculated with the Placzek's formula (Dhamelincourt et al., 1979). Raman intensity (integration area) was obtained using the specially designed Grams/32 software attached with the Renishaw System-1000 Laser Raman Spectrometer. The Raman cross-sections we used in our calculation for CO₂, CH₄ and N₂ contents are 1.21, 9.1 and 1.0, respectively. Based on laser Raman spectroscopic and microthermometric data of individual fluid inclusions, the bulk compositions were reconstructed with the method outlined in Ramboz et al. (1985) and are listed in Table 1. Any effect of salinity was ignored in the bulk fluid chemistry reconstruction because of the low salinity in the studied fluids. The critical parameter Z'_{CO₂} (Ramboz et al., 1985) used in our calculation was fixed at 2.0 mole% under the following considerations: (1) Micro-Raman spectrometric analysis at room temperature showed that the Raman peak of CH₄ was commonly at 2915 cm⁻¹± (2913 to 2917 cm⁻¹) equivalent to an inclusion internal pressure of

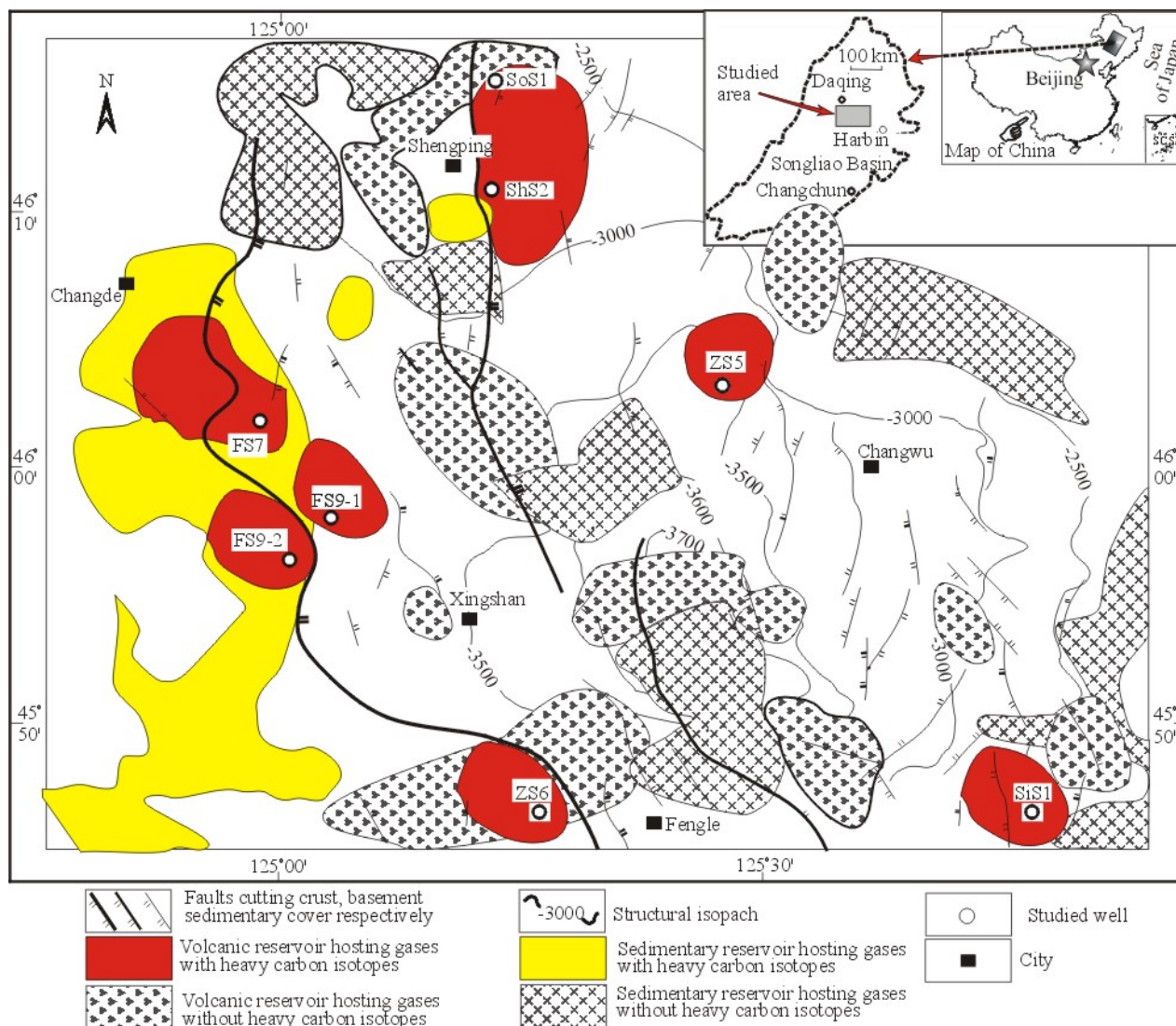


Fig. 1. Location and geological background of the studied wells in the central Songliao Basin, NE China, with a classification of commercial reservoirs.

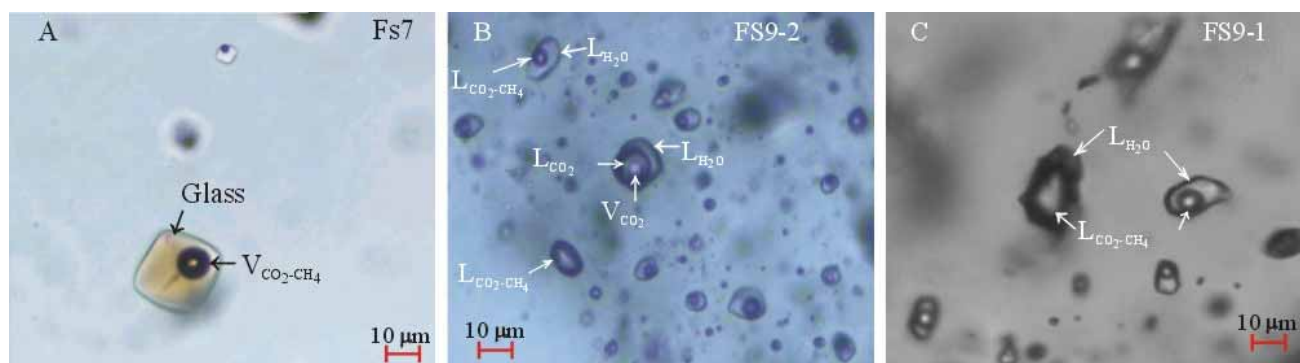


Fig. 2. Photographs of fluid inclusions from melt (A) and primary (B & C) sources in the reservoir volcanic rocks (rhyolite and dacite) of the Songliao Basin.

Sample numbers on the upper right correspond to those from Tables 1 and 2. V and L indicate phases of vapor and liquid.

ca. 100 bars according to Seitz et al. (1996) and, assuming an ideal mixture of real gases, the CO₂ solubility is about 1.06–2.2 mole% in low salinity water at the conditions of 10 MPa and from 0°C to room temperature (according the

experiment of Takenouchi and Kennedy, 1964); (2) Micro-Raman analysis for the liquid phase co-existing with a CH₄-CO₂ vapor showed that the dissolved CO₂ into the aqueous part of the inclusion is commonly in between 1.0–1.6

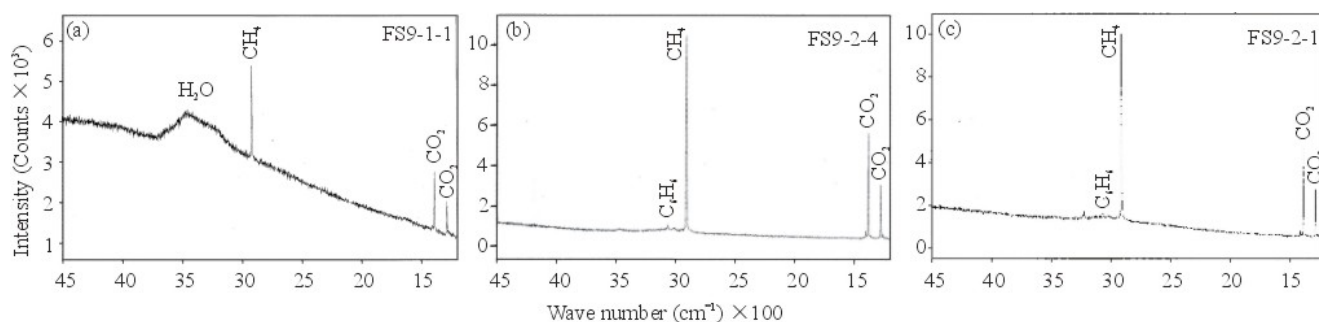


Fig. 3. Micro-Raman analyses of CH_4 - CO_2 - H_2O -bearing, vapor-dominated inclusions in the reservoir volcanic rocks of the Songliang Basin.

CH_4 in the samples is characterized by strong symmetric stretching lines at 2913 – 2917 cm^{-1} .

mole%.

4 Results and Discussions

4.1 Fluid inclusions

Fluid inclusions in matrix quartz or phenocryst/crystal quartz fragments are pervasive throughout the reservoir volcanic rocks. The inclusions occur both as secondary inclusions trapped along healed microfractures and as isolated or clustered primary inclusions that were trapped during mineral growth. Occasionally melts were trapped as glassy inclusions at high temperature (Fig. 2). In this study we focused on the latter two types of inclusions, especially the primary ones. Petrography, microthermometry and micro-Raman analysis indicate that quartz-hosted, CH_4 - CO_2 - H_2O -bearing fluid inclusions are the most common ones, followed by CH_4 - CO_2 monophase inclusions. Triple phase inclusions filled with vapor- CO_2 , liquid- CO_2 and liquid- H_2O can also be seen. Some inclusions contain a predominance of a CH_4 - CO_2 phase (Fig. 3b, c), which can simply be considered as a CH_4 - CO_2 binary system. For this reason we reconstructed only the CH_4 , CO_2 and H_2O concentrations in fluid bulk compositions taken from Table 1. Although such minor components as CO , N_2 , C_2H_2 , C_6H_6 , C_2H_4 , H_2S and H_2 have been frequently detected in many inclusions we did not list them in Table 1 because their concentrations are generally only around detection limits.

For each sample we have analyzed several tens of fluid inclusions and most of them are aqueous ones that give Raman spectra of a strong stretching band of water (3000 – 3700 cm^{-1}) and a small symmetric stretching band of methane ($2915\pm 2\text{ cm}^{-1}$) (Fig. 3a), which are similar to those described by Dubessy et al. (2001). High CH_4 fluid inclusions can occasionally be found. We selected one to five representative CH_4 / CO_2 -rich individual inclusions from each sample and list them here in Table 2. The size of the observed fluid inclusions ranged from 5 to $40\text{ }\mu\text{m}$ with a mode of 10 – $15\text{ }\mu\text{m}$. Their non-aqueous part occupied 30%

to 100% of the cavity volume equivalent to degrees of filling of the inclusion ranging between 0.7 to 0.0 (i.e. Flw , Ramboz et al., 1985) and most of them ranged from 30% to 50% ($Flw = 0.7$ – 0.5). Total homogenization temperature (Th) of the measured primary CH_4 / CO_2 -rich inclusions falls between 293 – 395°C with a mode of 305 – 315°C . The Th of the glassy inclusions is over 900°C . Melting temperature of solid CO_2 in the presence of vapor (Tm_{CO_2}) ranged from -56.9°C to -69.2°C with a mode around -61°C to -64°C . Homogenization temperature of the non-aqueous part to the liquid (Th_{CO_2}) is between -45.2°C and 8.6°C . Melting temperature of clathrate (Tm_{Clath}) ranged from 8.1°C to 14.8°C . Salinity of the fluid inclusions shown in Table 1 was derived from Tm_{Clath} using the method of Ridley and Hagemann (1999) or using the formula of Collins (1979) when the CH_4 content was low. The salinities in our case studies are all around or below 5% in NaCl equivalent weight percent.

4.2 Relationship between fluid inclusions and commercial gases hosted in the same volcanic rocks

The relationship of the reconstructed bulk compositions between CH_4 and CO_2 in the volcanic fluid inclusions is shown in Fig. 4. The resulting curve is composed of three segments: (1) the CO_2 -content of gases, which goes up rapidly with that of CH_4 in gas-poor inclusions; (2) the two components increase synchronously when $\text{CH}_4 > 10$ mole% and $\text{CO}_2 > 30$ mole%; (3) in gas-rich inclusions (more than 50 mole% CO_2 or CH_4), the CH_4 -content correlates negatively with that of CO_2 with a slope about -1 . This linear equation is nearly the same as that of bulk commercial gases hosted in the same volcanic reservoirs.

The former two cases seen in Fig. 4 relate to the fluid inclusions with relative high water content (79.3–88.4 mole%), in which concentrations of both CH_4 and CO_2 increase at the same time. Similar trends can be seen, as shown in Fig. 5, for the H_2O - CO_2 - CH_4 system in primary inclusions hosted in ferberite from the carbonaceous and aqueous stage in quartz (Ramboz et al., 1985). For

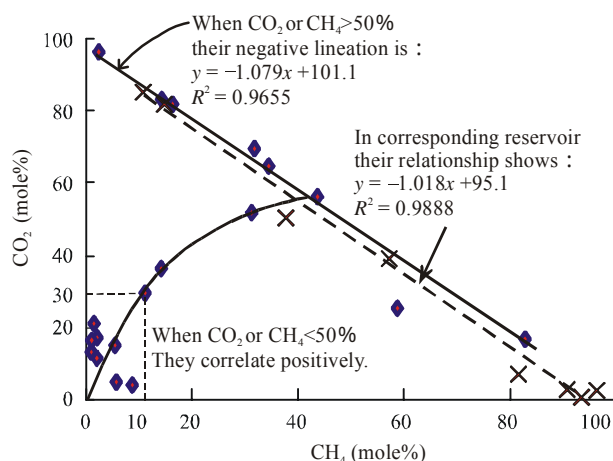


Fig. 4. Relationship of reconstructed bulk compositions between CO₂ and CH₄ in the primary inclusions of the volcanic rocks of the Songliao Basin (data from Table 1).

Dashed line and X indicate bulk gases in corresponding commercial reservoirs (data from Table 2).

comparison with the trapped gases in quartz we illustrate data of commercial gases trapped in the corresponding volcanic reservoirs (dashed line in Fig. 4). It is interesting to note that these commercial gases in the volcanic rocks also show strongly anti-correlated CH₄ and CO₂ (Fig. 4). This similarity tells us that CH₄ and CO₂ gases both from the fluid inclusions and from the reservoirs might have been regulated by similar processes. As we know that natural gas pools are fundamentally source-controlled, that is, a gas pool will mostly accumulate those gases with the most convenient migration path to the pool, we therefore interpreted those fluid inclusions with CH₄ or CO₂ more than 50% in mole as also source-controlled. There could be countless independent, inhomogeneously distributed sources of CH₄/CO₂ gases within the parent magma at the microscale, then the most convenient source of either CH₄ or CO₂ would contribute the most to the nearest inclusion and vice versa. This explanation can to some extent be proven by the inhomogeneous features of the fluid inclusions both in composition and distribution.

Trapped fluids in the volcanic rocks have a comparable composition to the content and isotopes of the gases from the same reservoirs. These affiliations can be attributed to their origin. The abiogenic hydrocarbons in the primary inclusions are initially mantle/magma derived. In addition, the inorganically derived gases, both CH₄ and CO₂, have contributed significantly to the corresponding gas pools. The results suggest that abiogenic methane can have industrial value and that the hydrocarbon budget of the bulk earth may be much larger than conventionally assumed.

4.3 Discussion

The Cretaceous geothermal gradient of the studied area

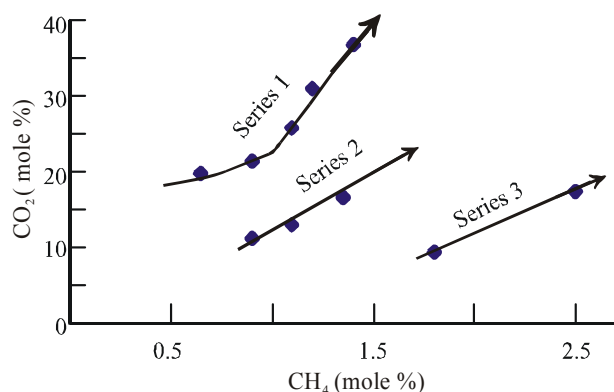


Fig. 5. Relationship of reconstructed bulk compositions between CO₂ and CH₄ in the primary inclusions hosted in ferberite from the carbonaceous and aqueous stages in quartz (data from Ramboz et al., 1985, Tables 2 and 3).

Note that CO₂ and CH₄ have a positive correlation for all the three series showing a similar trend to those low-concentration gases hosted also in the quartz of the reservoir volcanic rocks of the Songliao Basin. Series1: inclusions with relatively high homogenization temperature (10.8–18.2°C) of non-aqueous part; Series 2: water (81.8–87.5%)-rich inclusions; Series3: inclusions with relatively high melting temperature (10.2–10.7°C) of clathrate.

is from 27°C/km to 44°C/km (Wang et al., 1993). The higher value can be used to interpret the melt inclusion (*Th* 900°C) hosted in the volcanic quartz because it correlates with the magmatic activities of the basin evolution (Wang et al., 1993). Accordingly, the quotient of 900°C/44°C·km⁻¹ (ca. 20 km) can be considered as the maximum depth of inclusion formation because this would be the point of possible thermal equilibrium above which the magma might have been in complete thermal disequilibrium with the inclosing rocks during ascent. Taking into account that the Moho beneath the present Songliao Basin is at a depth of 30–35 km and that the boundary between the upper and lower crust is about 16 km (Yang et al., 2003), then a maximum closure depth of ca. 20 km suggests that the melt inclusions were formed at the lower crust depth or above.

The commercial gases hosted in CH₄-rich, fluid inclusion-bearing volcanic reservoirs (wells FS9-1 and FS9-2 in Table 2) show anomalous geochemical signatures with δ¹³C values of CH₄ and CO₂ varying between -27.11 to -27.25‰, and -4.06 to -5.46‰, respectively, and an R/R_a ratio ranging from 2.88 to 3.02. These signatures are comparable to those of the fumarolic gases from Nisyros and Veuvio (Fiebig et al., 2004), where CO₂ and CH₄ were originally mantle-derived. On the other hand, some accessory minerals contained in the reservoir volcanic rocks, such as magnetite and ilmenite, are important geocatalysts in the abiogenic formation of hydrocarbons (Horita, 2001). Moreover, abiogenic methane can be generated by different kind of reactions, such as water-graphite and water-carbonate interactions either in the crust or in the earth's interior at different pressures and

temperatures (Glassley, 1982; Takach et al., 1987; Berndt et al., 1996; Scott et al., 2004). Fischer-Tropsch reactions among volatiles and minerals can also produce methane (Potter et al., 2004). Logically, the CO₂-rich fluid hosted in the melt inclusion (e.g. sample FS7) demonstrates that within or below the lower crust existed CO₂-rich fluids originating from mantle/magma outgassing. Then mineral reactions involving conversion of Fe^{II} to Fe^{III} occurred during magma ascent and led to production of H₂ and conversion of dissolved CO₂ to reduced-C species including methane, some of which were then trapped as CH₄-rich fluid inclusions. These multiphase mineral-mineral-fluid interactions must have happened under thermal disequilibrium as a whole or under micro-scale equilibrium resulting in intense heterogeneity of fluids enclosed in the volcanic minerals.

5 Conclusions

(1) Fluids blocked in the primary inclusions of the volcanic reservoir rocks are synchronous volatiles with the lava that was originally mantle/magma derived. The results of this study demonstrate that these volatiles can yield high amounts of methane. The co-existence and positive correlation between CO₂ and CH₄ in the primary inclusions suggest that they came from the same deep-seated sources. We can therefore conclude that the detected CH₄-containing primary inclusions in the reservoir volcanic rocks should be considered as direct evidence for inorganically derived natural gas.

(2) Concentration of CH₄ in the fluid inclusions goes up with increasingly heavy carbon isotopes and in reversed order of $\delta^{13}\text{C}$ in alkane, implying a strong influence on isotopes of the reservoir gases. That is to say, mantle/magma-derived abiogenic alkanes must have contributed significantly to the gas pools.

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